

3.020 Lecture 22

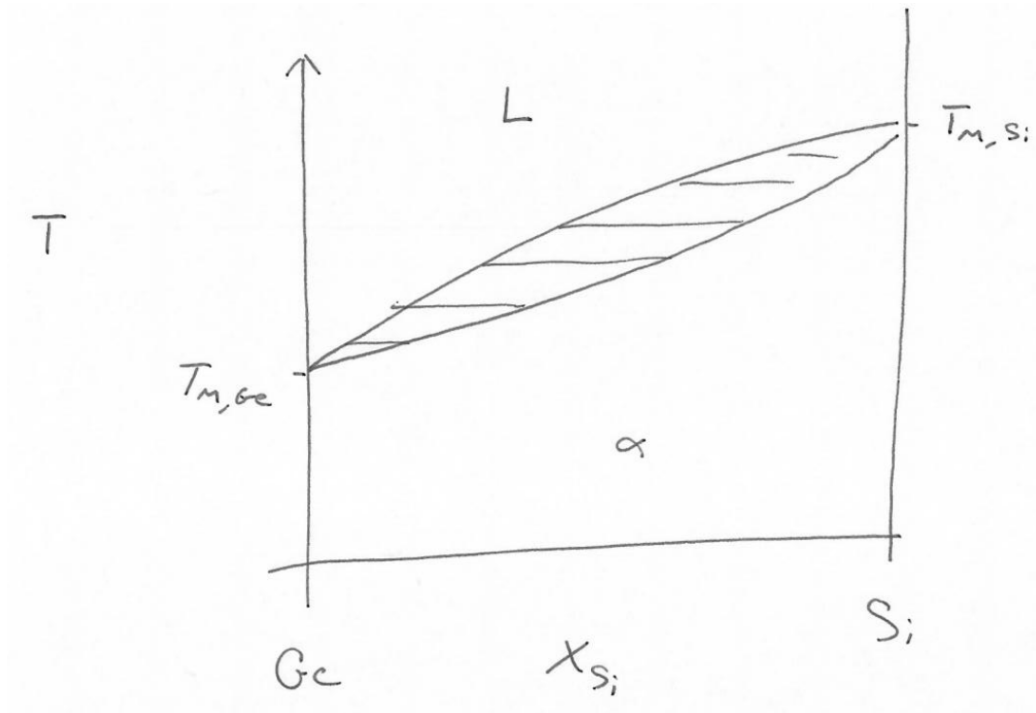
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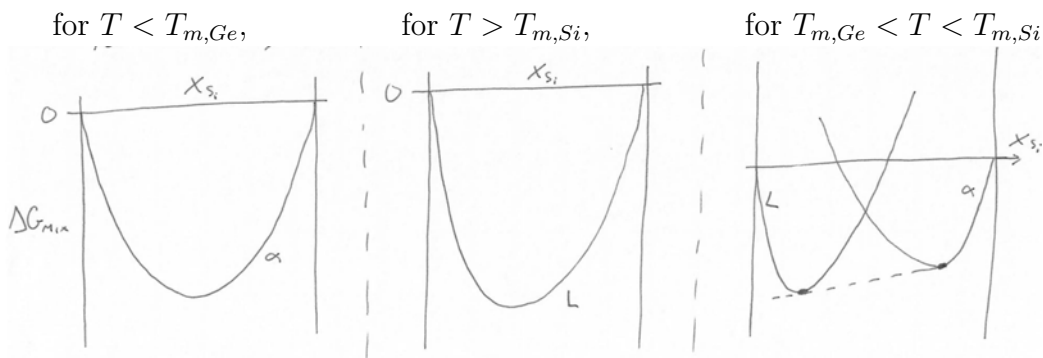
1 Thermodynamics of binary phase diagrams

- Thus far :
 - notation - bookkeeping, partial molar properties
 - solution modeling, ideal and simple regular models
 - spinodal systems: spontaneous unmixing when both pure components are in same structure

- More generally, pure components may be in different structures
e.g. Si-Ge system for $T_{m,Ge} < T < T_{m,Si}$

only need one
solu. model
in free energy-
composition
diagram



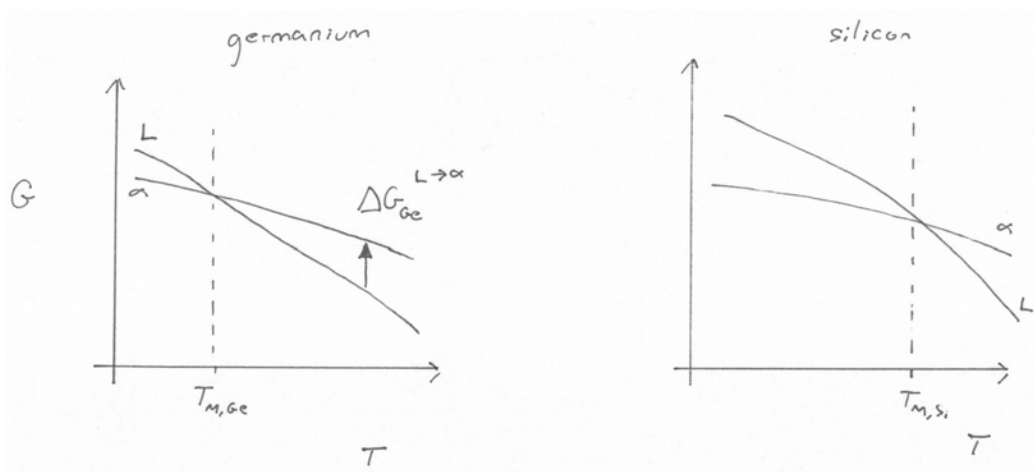


- Consider the process of making an α -phase Si-Ge solid solution at temperature $T_{m,Ge} < T < T_{m,Si}$

1. Convert the Ge from L to α phase _____
2. Mix solid Ge and Si in the α phase _____

this is a reference state change

- Temperature-dependence of free energy of pure components _____



this is described by solution modeling

from unary phase diagrams

- Accounting for $\Delta G_k^{\alpha \rightarrow \beta} = \Delta H_k^{\alpha \rightarrow \beta}(T) - T \Delta S_k^{\alpha \rightarrow \beta}(T)$
both terms may be temp-dependent
 - At equilibrium transition temp $T^{\alpha \rightarrow \beta}$, $\Delta G_k^{\alpha \rightarrow \beta} = 0$
 - If $\Delta C_P^{\alpha \rightarrow \beta} \approx 0$, then $\Delta H_k^{\alpha \rightarrow \beta}$ and $\Delta S_k^{\alpha \rightarrow \beta}$ are approximately temperature-independent, and the expression for the temp-dependence of $\Delta G_k^{\alpha \rightarrow \beta}$ simplifies:

$$\Delta G_k^{\alpha \rightarrow \beta}(T) \approx \Delta S_k^{\alpha \rightarrow \beta} \times (T^{\alpha \rightarrow \beta} - T)$$

- linear in T
- need data: transition temp $T^{\alpha \rightarrow \beta}$ and $\Delta S_K^{\alpha \rightarrow \beta}$

- Solving 2-phase coexistence condition for $T_{m,Ge} < T < T_{m,Si}$

1.

$$\mu_{Si}^{\alpha} = \mu_{Si}^L \quad \Longrightarrow \quad \mu_{Si}^0 + \Delta\mu_{Si,mix}^{\alpha} = \mu_{Si}^0 + \Delta\mu_{Si}^{\alpha \rightarrow L} + \Delta\mu_{Si,mix}^L$$

- μ_{Si}^0 : reference state of Si, which is α -phase
- $\Delta\mu_{Si,mix}^{\alpha}$: solution model for α -phase, PMP of mixing
- $\Delta\mu_{Si}^{\alpha \rightarrow L}$: reference state change
- $\Delta\mu_{Si,mix}^L$: solution model for L phase, PMP of mixing

2.

$$\mu_{Ge}^{\alpha} = \mu_{Ge}^L \quad \Longrightarrow \quad \mu_{Ge}^0 + \Delta\mu_{Ge}^{L \rightarrow \alpha} + \Delta\mu_{Ge,mix}^{\alpha} = \mu_{Ge}^0 + \Delta\mu_{Ge,mix}^L$$

- μ_{Ge}^0 : reference state of Ge, which is L-phase
- $\Delta\mu_{Ge}^{L \rightarrow \alpha}$: reference state change
- $\Delta\mu_{Ge,mix}^{\alpha}$: solution model for α -phase, PMP of mixing
- $\Delta\mu_{Ge,mix}^L$: solution model for L phase, PMP of mixing

- If both solutions behave ideally and if $\Delta C_P^{\alpha \rightarrow L} \approx 0$ for both Si and Ge, then coexistence condition and phase diagram can be solved explicitly, \rightarrow see DeHoff sec. 10.2.1, Fig. 10.22
- In all other cases we use computers \rightarrow CALPHAD

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